

BIOACTIVE CONSTITUENTS AND BIOCHEMICAL COMPOSITION OF THE EGYPTIAN BROWN ALGA *SARGASSUM SUBREPANDUM* (FORSK)

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ABSTRACT

During our search program for bioactive compounds from Egyptian marine sources, we isolated a spatane diterpene tetraol (**1**), fucosterol (**2**) and linoleic acid (**3**) from the brown alga *Sargassum subrepandum* (Forsk) C.Ag. In addition to those compounds, the four hydrocarbons were detected by using GC-MS named heptadecane, 2,6,10,14-tetramethyl-hexadecane, nonadecane and heneicosane. The chemical structure of **1** is assigned here definitely to first time by spectroscopic analyses including mass spectrometry (EI-MS, HR/EIMS), 1D and 2D NMR experiments. Phytochemical study of the unsaponified fraction of the algal extract and analysis by GC-MS confirmed the existence of nine compounds, among them 4,4,7a-trimethyl-5,6,7,7a-tetrahydrobenzofuran-2[4H]-one (**6**). In this article, we report as well the biochemical composition of *Sargassum subrepandum* (including ash, fibre, lipid, carbohydrate, total protein, and amino acids compositions) together with biological activities of its extract and the isolated compounds using diverse assays. www.relaquim.com

Keywords: Brown alga; *Sargassum subrepandum* (Forsk); Spatane diterpene; GC-MS; Biochemical composition; Biological activity

RESUMEN

Dentro de nuestro programa de compuestos bioactivos de fuentes marinas Egipcias, en este trabajo aislamos tetraol (**1**), fucosterol (**2**) y ácido linoleico (**3**) del alga café *Sargassum subrepandum* (Forsk) C.Ag. También detectamos por CG-EM cuatro hidrocarburos llamados heptadecano, 2,6,10,14-tetrametil-hexadecano, nonadecano y heniicosano. La estructura química definitiva de **1** se asignó aquí por primera vez por el análisis espectrométrico (EM-IE, AD/EMIE) y espectroscópico de RMN con experimentos de 1D y 2D. El estudio fitoquímico de la fracción no saponific-

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able del extracto de la alga y el análisis por CG-EM confirmó la existencia de nueve compuestos, entre ellos 4,4,7a-trimethyl-5,6,7,7a-tetrahydrobenzofuran-2(4H)-one (**6**). En este artículo también reportamos la composición bioquímica de *Sargassum subrepandum* (que incluye cenizas, fibra, lípidos, carbohidratos proteínas totales y composición de aminoácidos) y la actividad biológica de sus extractos y compuestos aislados utilizando diferentes ensayos. www.relaquim.com

Palabras clave: Brown alga; *Sargassum subrepandum* (Forsk); CG-EM; composición bioquímica; actividad biológica

INTRODUCTION

In the 1940s, natural product research interest was directed towards marine flora as potential sources in the production of metabolic constituents possessing antimicrobial, antiviral and antifungal properties (Baslow, 1969; Perez *et al.*, 1990), besides their effective pharmaceutical and economic properties (Ali & Pervez, 2003). Marine brown algae are prolific producers of interesting secondary metabolites, consisting of C₁₁-acetate derived compounds (Moore, 1978), sesquiterpenoids, diterpenoids (Gerwick *et al.*, 1981) and compounds of mixed biosynthesis origin (Fenical *et al.*, 1973). The diterpenoids from these algae are particularly unique, as the novel ring systems produced represent unconventional diterpenoid cyclizations, which have not yet observed from terrestrial sources (Gerwick *et al.*, 1980; Gerwick & Fenical, 1983).

Brown algae are also frequently encountered as the major vegetation in shallow-water tropical and subtropical habitats, even though herbivorous predators are plentiful. Hence, the correlation between secondary metabolite synthesis within this family and predator avoidance seems to be pronounced (Gerwick *et al.*, 1981). In the littoral zone of the Egyptian coast, brown algae are currently the most dominant group. Members of *Sargassum* genus represent valuable sources of a wide spectrum of complex lipids, essential fatty acids and

amino acids (Hossain *et al.*, 2003). *Sargassum subrepandum* (Forsk) C.Ag. is quite common in the Egyptian Red Sea coast (El-Naggar *et al.*, 1995), however, the bioactive constituents and biochemical composition of this alga have never been reported.

The marine brown alga *Sargassum subrepandum* (Forsk) was collected from Ras Gharib in the Egyptian Red Sea coast. The genus of *Sargassum* can be distinguished from its leaves and stem parts. Examination of the organic extract of the two parts of *Sargassum subrepandum* (Forsk) using TLC was visualized by spraying reagents, revealed the alga to contain a unique complex mixture of several components, including fatty acids, steroids and diterpenoids. Therefore, the whole alga was extracted with dichloromethane using a soxhlet. The afforded extract was then subjected for isolation using a series of chromatographic techniques. As a result, three major components were isolated named tetraol (**1**), fucosterol (**2**) (Popov *et al.*, 1985; Frost & Ward, 1968; Falsone *et al.*, 1994; Zhang *et al.*, 2007; Tang *et al.*, 2002; Sheu *et al.*, 1999) and linoleic acid (**3**) (Shaaban, 2004). On other hand, the less polar fractions were subjected to GC-MS, establishing the existence of heptadecane, 2,6,10,14-tetramethyl-hexadecane, nonadecane and heneicosane. Moreover, the biochemical composition of the alga was studied including the nutritional properties and the amino acids compositions of the alga.

The biological activity of the algal extract and its isolated compounds were as well evaluated against pathogenic microorganisms, and brine shrimp and tumor cell line MCF7 for cytotoxicity.

RESULTS AND DISCUSSION

ISOLATION AND STRUCTURE ELUCIDATION

Tetraol (**1**) was reported previously, however, with no proton-carbon assignments (Gerwick *et al.*, 1981). Therefore, we discuss the structure of **1** here in details, including a full assignment for its structure. Compound **1** was isolated as colourless oil and on TLC shown a weak UV absorption. The spot turned dark violet when treated with anisaldehyde/sulphuric acid spraying reagent. The molecular weight of **1** was deduced as 336 amu according to DCIMS, and the corresponding molecular formula, $C_{20}H_{32}O_4$, was proved by HRESIMS, possessing five degrees of unsaturation.

The 1H NMR/HMQC spectra of **1** showed two signals for doublet 1H at δ 5.79 (δ_c 137.7) of a *Trans*-olefinic proton ($J \sim 15.6$, 3.9 Hz), and its vicinal 1H as multiplet (δ 5.60, δ_c 131.9). Two 1H signals for an exo methylene (δ 5.30, 4.91, δ_c 108.2) and two oxy-methines were visible at δ 4.38 (δ_c 75.1) and δ 3.63 (δ_c 80.7). Signals for an oxy-methylene (δ_H 3.39, δ_c 70.6), five multiplet methines (δ_H 2.92~1.80; δ_c 44.7~37.8) and three methylenes (δ_H 2.28~1.70; δ_c 37.8~28.8) were observed. Three methyl signals were further visible, one of them was doublet at δ 0.93 (δ_c 14.9), while the remaining two methyls were singlets at δ 1.23 (δ_c 24.5) and 0.95 (δ_c 13.8).

The ^{13}C NMR/HMQC data of **1** displayed 20 carbon signals located in the sp^2 and sp^3 regions, including three C_q , among them one sp^2 (δ 151.1), one oxygenated sp^3 (δ 73.7), while the third was at δ 48.2. Base on the HMQC spectra, the respite 17 carbons

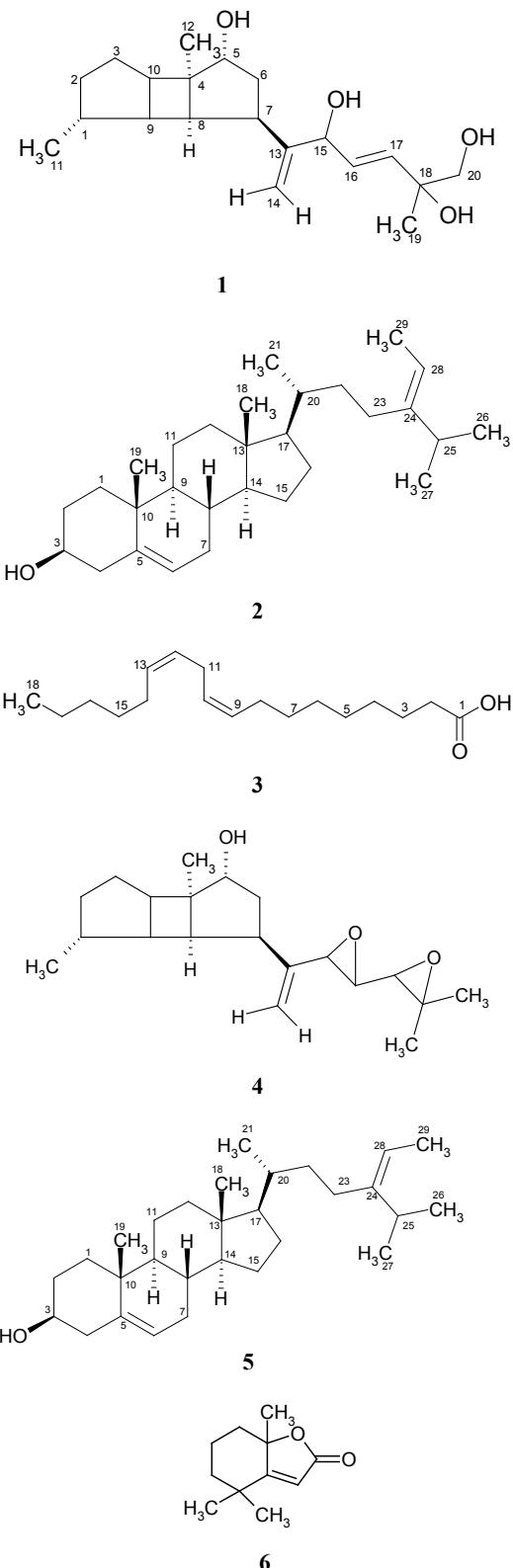


Figure 1. Chemical structures of the compounds obtained from *Sargassum subrepandum* (Forsk) C.Ag

were belonging to 9 CH, five methylenes and three methyl signals.

COSY analysis revealed the connectivity of H-17 (δ 5.79)/H-16 (δ 5.60), H-16 (δ 5.60)/H-15 (δ 4.38) as shown in Figure 2A. Another COSY analysis showed the connectivity of H-5 (δ 3.63)/H-6a or b (δ 2.28 or 1.66), H-6a or b (δ 2.28 or 1.66)/H-7 (δ 2.92), H-7 (δ 2.92)/H-8 (δ 1.96), H-8 (δ 1.96)/H-9 (δ 2.04), H-9 (δ 2.04)/H-10 (δ 1.96), H-10 (δ 1.96)/H-3a or b (δ 1.70 or 1.4), H-3 a or b (δ 1.70 or 1.4)/H-2 (δ 1.76), H-2 (δ 1.76)/H-1 (δ 1.80), H-1 (δ 1.80)/H-11 (δ 0.93), H-1 (δ 1.80)/H-9 (δ 1.80) as shown in figure 2. (δ_c 137.7) of a *Trans*-olefinic proton ($J \sim$ 15.6, 3.9 Hz), and its adjacent partner at δ 5.60 (δ_c 131.9) was fixed according to COSY experiment (Figure 2). Three additional signals were located at δ 5.30, 4.91 and 4.38, the first two of them were attributed to an exo methylene (δ_c 108.2), and the third was of an oxygenated methine (δ_c 75.1). The latter (δ 4.38) displayed a 3J COSY correlation with the olefinic methine at δ 5.60, recognizing their neighbourhood.

To assign the final structure of **1**, the HMBC experiment was used. In accordance, the exo-olefinic methylene protons H-14a and/or H-14b (δ 5.30 or 4.91) displayed three relevant correlations towards quaternary carbon C-13 (151.1), hydroxy-methine C-15 (75.1) and CH-7 (43.8) constructing an isobut-2-ene-1-ol moiety. The last fragment was in turn combined with the trans olefinic double bond *via* a mentioned COSY correlation between H-15 and H-16 beside to the clear HMBC correlation from H-17 (δ_H 5.79, δ_c 137.7) to C-15, and *vice versa*. Alternatively, the singlet methyl H₃-19 (δ 1.23) displayed three critical correlations; two among them were towards C-17 and the oxy-methylene C-20 (δ 70.6), beside to its directly attached C-18 (δ 73.7). This assignment was further recognized as the H₂-20 showed two correlations at C-17 and C-18, respectively, confirming the partial structure **A**.

In addition to exhibited COSY coupling between the doublet methyl H₃-11 (δ 0.93) and H-1 (δ 1.80), this methyl showed rather three HMBC correlations at CH-1 (δ 37.8), CH₂-2 (δ 36.3) and CH-9 (δ 38.7). The methylene CH₂-2 (δ 1.76) displayed two HMBC correlations to CH-10 (δ 44.1) and CH-9 (δ 38.7), beside to a visible COSY at H-3a (δ 1.70 and H-3b (δ 1.40), recognizing a 1-methyl-2,3-disubstituted cyclopentane **B**. Proton signal (δ 0.95) of the third methyl CH₃-12 exhibited four further correlations at C-4 (δ 48.2), CH-10 (δ 44.1), CH-8 (δ 44.7) and the oxy methine C-5 (δ 80.7). The last methine CH-8 (δ 1.96) was confirmed to be directly connect with CH-9 (δ 2.04) according to a shown COSY correlation. In the HMBC spectra, the methylene protons H-3_{a,b} ((δ 1.70, 1.40) displayed correlation to C-4 (δ 48.2) confirming a fusion between cyclobutane and ring B *via* C-4 and CH-8, affording fragment C. According to COSY, H-5/H₂-6/H-7, were confirmed to be directly attached, constructing a spatane moiety as shown in partial structure **A** (Figure 2).

In accordance, the final structure of **1** was recognized as (*E*)-6-(3-Hydroxy-3a,6-dimethyl-decahydro-cyclobutadicyclopenten-1-yl)-2-methyl-hepta-3,6-diene-1,2,5-triol; tetraol; a spatane diterpene (Figure 3). As the insufficiency of the compound's (**1**) amount, it was not able to fix its absolute configuration. Spatanes are 5-4-5 membered tricyclic diterpenes, which have not been reported from terrestrial sources. Spatol (**4**) the first member of this class of diterpenes, was isolated from the brown alga *Spatoglossum schmittii* (Gerwick *et al.*, 1980; Venkateswarlu & Biabani, 1995) and subsequently from *Stoechospermum maroinatum* (Gerwick *et al.*, 1981; Rao *et al.*, 1987). Some of this class of diterpenes showed cytotoxic (Gerwick *et al.*, 1980) and antibacterial (De Silva *et al.*, 1982) activities.

The unpolar fractions of the algal extract were deduced to contain several bands of extremely confused components; however,

it was not able to purify them by the usual techniques. So, they were being applied to detection on the bases of GC-MS. In accordance, four hydrocarbons were fixed; heptadecane ($R_t = 18.30$ min) (Acevedo *et al.*, 2010), 2,6,10,14-tetramethyl-hexadecane ($R_t = 18.37$ min) (Watanabe *et al.*, 2008), nonadecane ($R_t = 20.33$ min) (Meng *et al.*, 2008) and heneicosane ($R_t = 22.19$ min) (Jerkovicet & Marijanovic, 2009).

Lipids form large group of natural compounds, which are water insoluble, but very soluble in hydrocarbons and ether. Lipids are mostly glyceride esters of long chain carboxylic acid, fatty acids, which

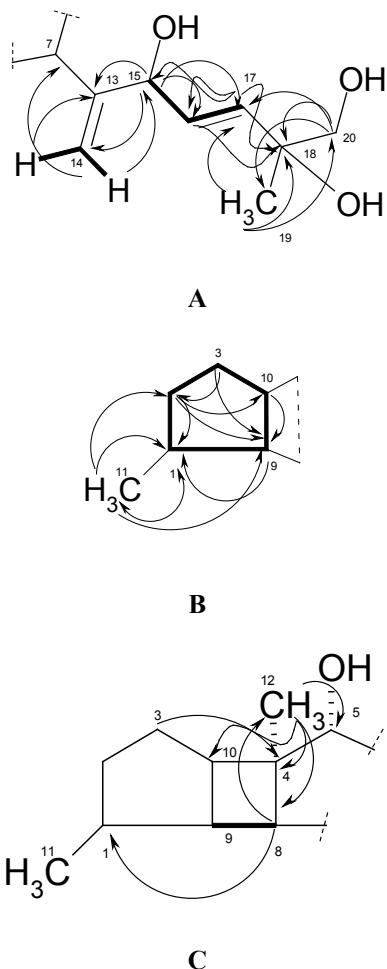


Figure 2. H,H COSY (—) and selective HMBC (→) correlations of the partial structures A, B and C of tetraol (1)

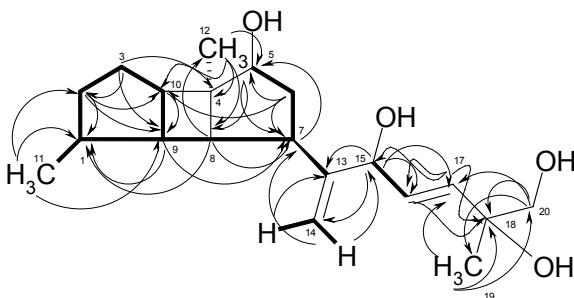


Figure 3. H,H COSY (—) and selective HMBC(→) connectivities of tetraol (1)

are usually unbranched, oils and fats. The most important members of this class are widely distributed throughout animal and vegetable kingdom. An application of the ethereal extract of the desired marine alga to saponification and working up, the obtained unsaponified fraction from *Sargassum subrepandum* (0.05 g, 0.5 %) was estimated on the bases of GC-MS analyses. In accordance, nine hydrocarbons were detected, 2-ethyl-1-hexanol ($R_t = 8.13$), 4,4,7a-trimethyl-5,6,7,7a-tetrahydrobenzofuran-2(4H)-one (**6**) ($R_t = 15.53$), heptadecane ($R_t = 17.20$), nonadecane ($R_t = 18.28$), 2,6,10,14-tetramethyl-hexadecane ($R_t = 18.35$), 4,6,10-trimethyl-2-pentadecanone ($R_t = 18.76$), 6,10,14-trimethylpentadecan-2-ol ($R_t = 18.83$), heneicosane ($R_t = 20.32$) and heptacosane ($R_t = 21.27$).

CHEMICAL COMPOSITION OF *SARGASSUM SUBREPANDUM*

Nutritional properties of *Sargassum subrepandum*

Generally, the nutritional properties of seaweeds are usually determined from their biochemical composition such as protein, carbohydrates, crude fibre, lipids (fats) and ash content (McDermid & Stuercke, 2003). Biochemical composition of *Sargassum subrepandum* is shown in Table 1. Ash content was the most abundant component of dried material in *Sargassum subrepandum*.

dum (29.49 %) as similar to those reported (Matanjun *et al.*, 2009).

From Table 1, it is apparent that crude fibre composition in *Sargassum subrepandum* had markedly high content (7%) (Rupérez & Saura-Calixto, 2001). The fibre content of seaweed varieties is higher than those found in most fruits and vegetables. The human consumption of algal fibre has been proven to be health-promoting and its benefits (Dawczynski *et al.*, 2007). Dietary fibre consumption promotes the growth and protection of the beneficial intestinal flora. Moreover, its consumption, in combination with high glycemic load foods reduces the overall glycemic response, greatly increases stool volume and reduces the risk of colon cancer (Goni *et al.*, 2000; Guidel-Urbano & Goni, 2002; Jiménez-Escrig & Sánchez-Muniz, 2000).

Lipid content of marine macroalgae accounts for 1-6 g/100g d.w. (Fleurence *et al.*, 1994; Herbreteau *et al.*, 1997). The lipid content of *Sargassum subrepandum* was presently of 3.61 %. Carbohydrate contents in *S. subrepandum* represented a ratio of 10.21 %. Burtin (2003) stated that most of those seaweeds contain large amounts of polysaccharides, which are mostly concentrated in algal cell walls, e.g. alginates predominate in brown seaweeds. Burtin (2003) revealed that the protein content of brown seaweeds is generally small (average: 5-15 % of the dry weight). In *Sargassum subrepandum*, the examined total protein

content her was found to be 3.2 % (Table 1). This low ratio of protein content is mostly seasonal affected (Galland-Irmouli, 1999).

Amino acid composition of *Sargassum subrepandum*

Concentrations of protein amino acids in *Sargassum subrepandum* are listed in Table 2. In accordance, glutamic acid (8.86 mg/g), valine (7.89 mg/g) and aspartic acid (7.44 mg/g) represent the most predominant amino acids in this alga. Furthermore, the results indicated that methionine was present in low concentration (1.79 mg/g). Moreover, a total amount of the protein amino acids represent a large amount in *Sargassum subrepandum* (78.99 mg/g), indicating the high tendency of these taxa (species) to accumulate the amino acids in their tissues.

BIOLOGICAL ACTIVITIES

Patterns of activity against set of microorganisms namely; *Bacillus subtilis*, *Staphylococcus aureus*, *Streptomyces viridochromogenes* (Tü 57), *Escherichia coli*, *Candida albicans*, *Mucor miehi*, *Chlorella vulgaris*, *Chlorella sorokiniana*, *Scenedesmus subspicatus*, *Rhizoctonia solani* and *Pythium ultimum* have been carried out for crude extract of *S. subrepandum* (Forsk.) C.Ag. (40 microg/disc, Ø 9 mm) indicating its moderate activity against the Gram positive *Staphylococcus aureus* (14 mm). Moreover, the algal extract was found to reveal a potent antitumor activity at IC₅₀ of 0.94 microg/mL against the breast carcinoma tumor cell line MCF7 in comparison with doxorubicin (IC₅₀ 0.7 microg/mL). In contrast, the extract exhibited a very weak cytotoxic activity against the brine shrimp (1.7 %, 100 microg/mL). Inversely, compounds **1-3** showed no cytotoxic or antimicrobial activity against any of the mentioned test organisms, indicating that

Table 1. Biochemical composition of *S. subrepandum* (values expressed as % dry weight).

Composition	%
Ash	29.49 ± 3.37
Crude fiber	7 ± 0.27
Crude lipid	3.61 ± 0.05
Soluble CHO ¹	10.21 ± 0.70
Total protein	3.2 ± 0.12

¹CHO = Carbohydrate.

²Data are means of three determinations ± SD.

Table 2. The protein amino acid profile of *S. subrepandum* (values expressed as mg g⁻¹ dry weight).

Amino acid	mg g ⁻¹	Amino acid	mg g ⁻¹
Aspartic acid	7.44 ± 0.11	Isoleucine	4.14 ± 0.22
Threonine	4.47 ± 0.10	Leucine	5.93 ± 0.55
Serine	4.60 ± 0.08	Tyrosine	2.09 ± 0.30
Glutamic acid	8.86 ± 0.12	Phenylalanine	3.14 ± 0.69
Proline	2.00 ± 0.07	Histidine	7.45 ± 0.56
Glycine	5.03 ± 0.92	Lysine	4.41 ± 0.92
Alanine	6.68 ± 0.14	Arginine	3.07 ± 0.46
Valine	7.89 ± 0.06	TEA ¹	42.29 ± 3.61
Methionine	1.79 ± 0.05	TAA ²	78.99 ± 5.35

¹Total essential amino acids.²Total amino acids.

the activity of the crude extract was due to some minor components which was not be able to isolate in sufficient amounts for analysis.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES

The NMR spectra were measured on a Bruker AMX 300 (300.135 MHz), a Varian Unity 300 (300.145 MHz) and a Varian Inova 600 (150.820 MHz) spectrometer. ESI MS was recorded on a Finnigan LCQ with quaternary pump Rheos 4000 (Flux Instrument). EI mass spectra were recorded on a Finnigan MAT 95 spectrometer (70 eV). DCI mass spectra were recorded on Finnigan MAT 95 A (200 eV) using reactant gas NH₃. GC-MS was used a Trace GC-MS Thermo Finnigan, ionization mode EI eV 70, instrument equipped with a capillary column CP-Sil 8 CB for amines (length: 30 m; inside diameter: 0.25 mm; outside diameter: 0.35 mm; film thickness: 0.25 μm). The analysis was carried out at a programmed temperature: initial temperature 40°C (Kept for 1 min), then increasing at a rate of 10°C/min and final temp 280 °C (kept for 10 min), Injector temp was 250 °C and detector (mode of ionization: EI) temp at 250 °C, He as a carrier gas at flow rate 1

mL/min, total run time 27 min and Injection volume 0.2 μL. Flash chromatography was carried out on silica gel (230-400 mesh). R_f-values were measured on Polygram SIL G/UV₂₅₄ (Macherey-Nagel & Co.). Size exclusion chromatography was done on Sephadex LH-20 (Pharmacia).

COLLECTION AND TAXONOMY OF THE MARINE ALGA

The alga *S. subrepandum* (Forsk.) C.Ag. was collected from Ras Gharib on Suez-Gulf, Red Sea, Egypt. The identification was carried out according to method reported previously (Nasr, 1940; Abou-ElWafa, 2005). A voucher reference specimen (no. 4200) of the alga is deposited at the Department of Botany, Faculty of Science, Mansoura University, Egypt.

PREPARATION OF SAMPLES

The collected samples were cleaned up from epiphytes and non-living matrix in running water, and rinsed many times in distilled water. The samples were then spread on string nets and allowed to dry in air. The air-dried samples were ground and stored in suitable closed bottles at room temperature.

EXTRACTION AND ISOLATION OF THE ACTIVE CONSTITUENTS

The air dried leaves (c.a. 100 g) and stems (c.a. 30 g) of *S. subrepandum* were individually applied to extraction by soxhlet using CH_2Cl_2 for ~ 12 hrs. The afforded extracts were filtered off and concentrated *in vacuo* at 40°C to yield greenish brown crude extracts (2.24 g and 0.37 g) from leaves and stems, respectively. As the identical TLC for both extracts, they were combined (2.61 g) and applied to Sephadex LH-20 (MeOH). TLC monitoring, visualized under UV light and spraying with anisaldehyde/sulphuric acid furnished three fractions. Fraction I (1.5 g) was mostly constructed from undesired pigments, including chlorophyll and fatty acids. The fraction was applied to silica gel column eluted with C_6H_{12} - CH_2Cl_2 gradient, and the delivered unpolar components were combined and submitted to GC-MS analysis. Fraction II (0.71 g) was applied to Sephadex LH-20 column (CH_2Cl_2 /40% MeOH) lead to three sub-fractions, FIIa (0.29 g), FIIb (0.18 g), and FIIc (0.28 g). Sub-fraction FIIa was applied to silica gel column (C_6H_{12} - CH_2Cl_2) giving unpolar mixture of oil, KSSR5 (120 mg), which was detected by GC-MS analysis. The two sub-fractions FIIa and FIIb were combined (0.46 g) and applied to silica gel column (CH_2Cl_2 -MeOH) affording two components, fucosterol (**2**, 27 mg) and linoleic acid (**3**, 18 mg). FIII (0.39 g) was purified on silica gel column (CH_2Cl_2 - CH_3OH) to obtain a colourless oil of **1** (3 mg).

Tetraol; 5(R),15,18(R),19-Tetrahydroxy-penta-13,16(E)-diene (**1**): $\text{C}_{20}\text{H}_{32}\text{O}_4$ (336); an UV faint absorbing colourless oil, turned dark violet with anisaldehyde/sulphuric acid and heating. – DCI MS (NH_4): m/z = 354 ($[\text{M}+\text{NH}_4]^+$). – HRESI MS: 359.21921 (calc. 359.21927 for $\text{C}_{20}\text{H}_{32}\text{O}_4\text{Na}$). – ^1H NMR (CDCl_3 , 300 MHz), δ (ppm): 1.80 (m, 1H, H-1), 1.76 (m, 1H, H-2), 1.70 (m, 1H, H-3_a), 1.40 (m, 1H, H-3_b), 3.63 (d, J = 4.1 Hz, 1H, H-5), 2.28 (ddd, J = 13.0, 13.0, 4.3

Hz, 1H, H_a-6), 1.66 (m, 1H, H_b-6), 2.92 (m, 1H, H-7), 1.96 (m, 1H, H-8), 2.04 (m, 1H, H-9), 1.96 (m, 1H, H-10), 0.93 (d, J = 6.0 Hz, 3H, H₃-11), 0.95 (s, 3H, H₃-12), 5.30 (d, J = 1.3 Hz, 1H, H_a-14), 4.91 (s, 1H, H_b-14), 4.38 (d, J = 8.6 Hz, 1H, H-15), 5.60 (m, 1H, H-16), 5.79 (dd, J = 15.6, 3.9 Hz, 1H, H-17), 1.23 (s, 3H, H₃-19), 3.39 (s, 2H, H₂-20). – ^{13}C NMR (CDCl_3 , 150 MHz), δ (ppm): 37.8 (CH-1), 36.3 (CH₂-2), 28.8 (CH₂-3), 48.2 (C_q-4), 80.7 (CH-5), 37.8 (CH₂-6), 43.8 (CH-7), 44.7 (CH-8), 38.7 (CH-9), 44.1 (CH-10), 14.9 (CH₃-11), 13.8 (CH₃-12), 151.1 (C_q-13), 108.2 (CH₂-14), 75.1 (CH-15), 131.9 (CH-16), 137.7 (CH-17), 73.7 (C_q-18), 24.5 (CH₃-19), 70.6 (CH₂-20).

Fucosterol (**2**): $\text{C}_{29}\text{H}_{48}\text{O}_1$ (412); Colourless solid, turned blue on spraying with anisaldehyde/sulphuric acid and heating. – EI MS: m/z (%) = 413 ($[\text{M}]^+$, 28), 399 ($[\text{M}-\text{CH}_2]^+$, 10), 314 (100), 299 (24), 271 (10), 255 (8), 229 (11), 211 (7), 145 (8), 119 (9), 81 (24), 69 (36), 55 (70), 44 (39), 41 (28). – ^1H NMR (CDCl_3 , 300 MHz), δ (ppm): 5.37 (d, J = 4.7 Hz, 1H, H-6), 5.19 (q, 1H, J = 6.3 Hz, H-28), 3.51 (m, 1H, H-3), 2.27 (m, 1H, H-25), 2.22 (m, 1H, Ha-4), 2.10-1.83 (m, 4H, H₂-7, H₂-22), 1.58 (d, 3H, J = 6.1 Hz, CH₃-29), 1.55-1.18 (m, 6H), 1.15-1.05 (m, 6H), 1.01 (s, 3H, CH₃-19), 0.99 (6H, CH₃-26, CH₃-27), 0.94 (d, 3H, J = 6.1 Hz, CH₃-21), 0.93-0.80 (m, 8H), 0.68 (s, 3H, CH₃-18). – ^{13}C NMR (CDCl_3 , 125 MHz), δ (ppm): 36.5 (CH₂-1), 31.7 (CH₂-2), 71.8 (CH-3), 42.4 (CH₂-4), 140.7 (C_q-5), 121.6 (CH-6), 31.9 (CH₂-7), 31.9 (CH-8), 50.1 (H-9), 37.3 (C_q-10), 21.1 (CH₂-11), 28.3 (CH₂-12), 42.3 (C_q-13), 56.8 (CH-14), 24.4 (CH₂-15), 39.8 (CH₂-16), 55.8 (H-17), 13.2 (CH₃-18), 19.5 (CH₃-19), 36.4 (CH-20), 18.8 (CH₃-21), 35.3 (CH₂-22), 25.7 (CH₂-23), 146.9 (C_q-24), 34.8 (CH-25), 22.2 (CH₃-26), 22.3 (CH₃-27), 115.5 (CH-28), 11.9 (CH₃-29).

Linoleic acid; (9Z,12Z)-9, 12-octadecanoic acid (**3**): $\text{C}_{18}\text{H}_{32}\text{O}_2$ (280.45); an UV absorbing colourless oil, turned to blue by anisaldehyde/sulphuric acid and heating.

– EI MS (70 eV): m/z (%) = 280 (80), 264

(28), 137 (10), 124 (15), 110 (28), 95 (60), 81 (84), 67 (100), 55 (92), 41 (92). – ^1H NMR (CDCl_3 , 300 MHz), δ (ppm): 8.98 (s, br, 1 H, COOH), 5.43-5.28 (m, 4 H, 9,10,12,13-CH), 2.78 (t, $^3J = 6.0$ Hz, 2 H, 11- CH_2), 2.38 (t, $^3J = 7.2$ Hz, 2 H, 2- CH_2), 2.08 (m, 4 H, 8,14- CH_2), 1.63 (m, 2 H, 3- CH_2), 1.42-1.23 (m, 14 H, 4,5,6,7,16,17-CH₂), 0.85 (m, 3 H, 18- CH_3). – ^{13}C /APT NMR (CDCl_3 , 50 MHz), δ (ppm): 180.1 (CO, C_q), 130.1 (CH-13), 129.9 (CH-9), 128.0 (CH-10), 127.8 (CH-12), 31.5 (CH₂-2), 29.6 (CH₂-16), 29.6 (CH₂-11), 29.5 (CH₂-14), 29.3 (CH₂-8), 29.1 (CH₂-7), 29.0 (CH₂-6), 29.0 (CH₂-5), 27.1 (CH₂-4), 25.6 (CH₂-3), 24.7 (CH₂-15), 22.5 (CH₂-17) 14.0 (CH₂-18).

Phytosterols and hydrocarbons estimation

A powdered sample (10 g) of *S. subrepandum* was extracted with petroleum ether (60-80 °C) at room temperature. The afforded extract was concentrated *in vacuo* to give an oily residue, which was then treated with 50 mL of 10 % alcoholic KOH and refluxed in a water bath for 2 hrs. After cooling, 50 mL water was then added and the aqueous solution was extracted with chloroform. The organic layer was washed with water until it became alkali free, and dried over anhydrous Na_2SO_4 . The solvent was distilled off to give the unsaponified fraction as an oily extract. Finally, the oily sample was applied to GC-MS analysis, and the retention time of their peaks was compared with reported data in literatures.

Chemical composition of *S. Subrepandum*

Ash content

The ash content of the alga was estimated according to Marsham, Scott & Tobin (2007). Four grams of dried algal material were added to a preweighed crucible. The algal contents were then inserted in a muffle furnace (Metrawatt GmbH, RO-8) at 550 °C for ~10 hrs. After cooling,

the contents were kept in a desiccator (to avoid any atmospheric moisture) and reweighed.

Lipid content

Ten grams from the air dried alga were applied to extraction with petroleum ether 40-60 °C during Soxhlet extractor (Soxtec System HT6, Tecator, Hoganas, Sweden) for ~10 hrs. After concentration *in vacuo*, the obtained crude lipids were weighed gravimetrically according to Wong and Cheung (2000).

Protein content

0.5 Gram from the air dried alga was extracted with 10 mL phosphate buffered saline (pH = 7). After centrifugation, 0.3 mL of the supernatant was raised to 1 mL by addition of phosphate buffered saline and treated with 5 mL coomassie brilliant blue- G250. According to the Bradford method (1976), the afforded sample was then applied to colorimetric measurements at wavelength 595 nm in comparison with bovine serum albumin (BSA) as reference. This was carried out using ATI UNICAM UV/Vis Spectrometer UV.2 (England).

Carbohydrate content

Soluble carbohydrates were determined according to the colorimetric method of Yemm and Willis (1954). Known weight of dried algal sample (0.1 g) was hydrolyzed with 5 mL 2.5 N HCl for 3 hrs in a boiling water bath, cooling and neutralization with Na_2CO_3 till effervescence ceases. After neutralization; the volume was completed to 100 mL using distilled water and filtered. 0.5 mL of the supernatant was completed to 1 mL using distilled water and mixed with 4 mL of 0.1 % anthrone/sulphuric acid and heated for 10 min in a water bath. After cooling, the developed blue green colour was measured at 630 nm using a spectrophotometer ATI

UNICAM UV/VIS Spectrometer UV.2 (England). Carbohydrate concentration was calculated in comparison with D-glucose as a reference sugar.

Crude fibre content

Crude fibre was estimated by acid and alkaline digestion. Two grams of defatted residue was boiled with 200 mL of 1.25% H_2SO_4 for 30 min, cooled, filtered and the residue was washed (3 ×) with 50 mL of boiling water. The washed residue was digested by boiling in 200 mL of 1.25% NaOH for 30 min, cooled, filtered and washed (3 ×) with 50 mL of boiling water and finally with 25 mL ethanol. The residue was dried in an oven at 110 °C to constant weight using a pre-weighed crucible, cooled in desiccator and weighed. The residue was ashed at 550 °C for 8 hrs, cooled in a desiccator and reweighed. Crude fibre content was expressed as percentage loss in weight on ignition (Nesamvuni, Steyn & Potgieter, 2001; AOAC, 2000).

Amino acid content

Three grams of dried seaweed sample were treated with 6N HCl containing one crystal of phenol (approx. 0.5 mg) in a pyrex tube (6 × 50 mm) and (100 µL) of 2-mercaptoethanol. After sealing the vial under vacuum, the reaction mixture was applied to hydrolysis at 110 °C for 24 hrs. The vial was then applied to evaporation *in vacuo*, and the residue was dissolved in methanol and applied for subsequent chromatographic separation. Amino acid mixture in the hydrolysates was determined by an amino acid analyzer LC 3000 eppendorf/Biotronik using column type H 125 × 4 mm, pre-column type H 60 × 4 mm, eluents and reagent type H1 (4 buffer system) (Walker, 1996). Determination of tryptophan was carried out using method described by Miller (1967) after hydrolysis of samples with barium hydroxide.

BIOLOGICAL ACTIVITY

Antimicrobial activity

Antimicrobial assays were conducted utilizing the disc-agar method (Burkholder, Burkholder & Almodovar, 1960) against diverse sets of microorganisms. The *Sargassum subrepandum* extract was dissolved in CH_2Cl_2 /10% MeOH at a concentration of 1 mg/mL. Aliquots of 40 µL were soaked on filter paper discs (9 mm, no. 2668, Schleicher & Schüll, Germany) and dried for 1 h at room temperature under sterilized conditions. The paper discs were placed on inoculated agar plats and incubated for 24 hrs at 38 °C for bacterial and 48 hrs (30 °C) for the fungal isolates, while the algal test strains were incubated at ~ 22 °C in day light for 8~10 days. The algal extract was examined against the following test microorganisms: *Bacillus subtilis*, *Staphylococcus aureus*, *Streptomyces viridochromogenes* (Tü 57), *Escherichia coli*, *Candida albicans*, *Mucor miehi*, *Chlorella vulgaris*, *Chlorella sorokiniana*, *Scenedesmus subspicatus*, *Rhizoctonia solani* and *Pythium ultimum*.

Brine shrimp microwell cytotoxic assay

The cytotoxic assay was performed according to Takahashi *et al.* method (Takahashi, Kurasawa, Ikeda, Okami & Takeuchi, 1989) and Sajid *et al.* screening (Sajid, Fondja Yao, Shaaban, Hasnain & Laatsch, 2009).

In vitro cytotoxic activity using SRB assay

In vitro cytotoxicity of the desired components were tested according to (Skehan, Storeng, Scudiero, Monks, McMahon, Vistica, Warren, Bokesch, Kenney & Boyd, 1990). Cells were plated in a 96-multiwell plate (104cells/well) for 24 hrs before treatment with the components to be tested to allow attachment of cells to the wall of the plate. Different concentrations of the components under test (0, 1, 2.5, 5, 10 µg/

mL) were carried out and added to the cell monolayer. Triplicate wells were prepared for each individual dose. Monolayer cells were incubated with tested components for 48 hrs at 37 °C in an atmosphere with 5% CO₂. After incubation, cells were fixed, washed and stained with Sulfo-Rhodamine-B pigment. Excess of the dye was washed off with acetic acid, while the adsorbed dye was dissolved with Tris-EDTA buffer. Colour intensity was measured in an ELISA reader. The relation between surviving fraction and drug conc. was plotted to get the survival

curve of each tumor cell line after application of the specified compound.

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